

# Carbohydrates: Interface between Chemistry and Biology

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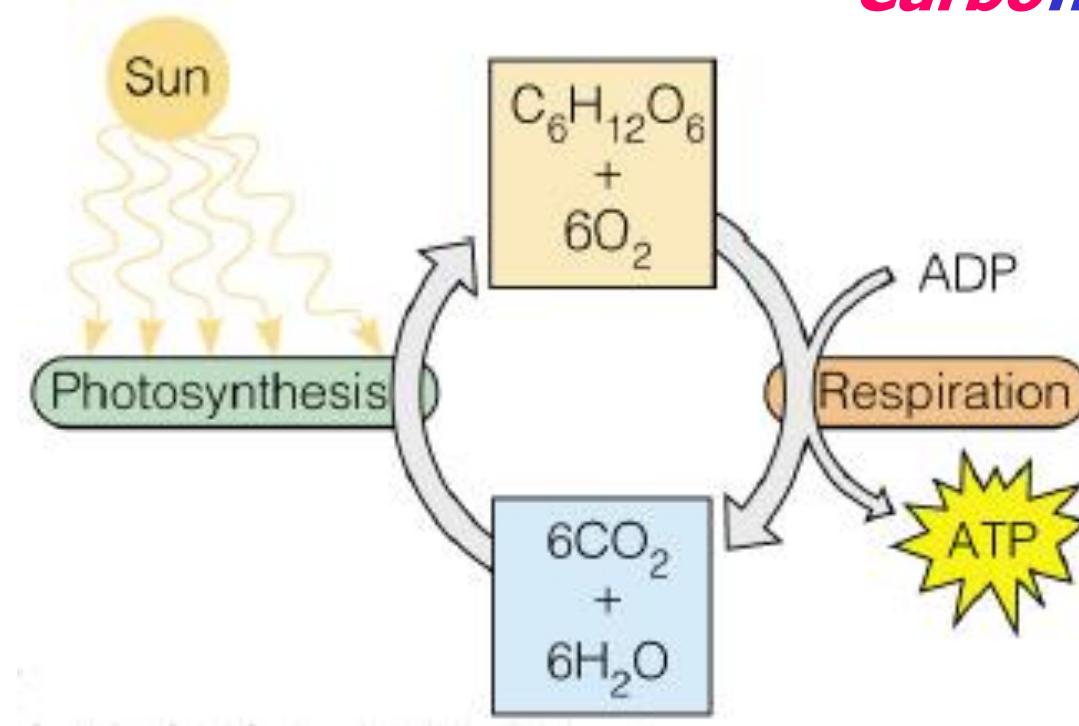
Maharana Pratap Govt. P.G. College, Hardoi

# What is carbohydrate

General formula:  $C_nH_{2n}O_n$

$C\ H_2O$

*Carbohydrate*



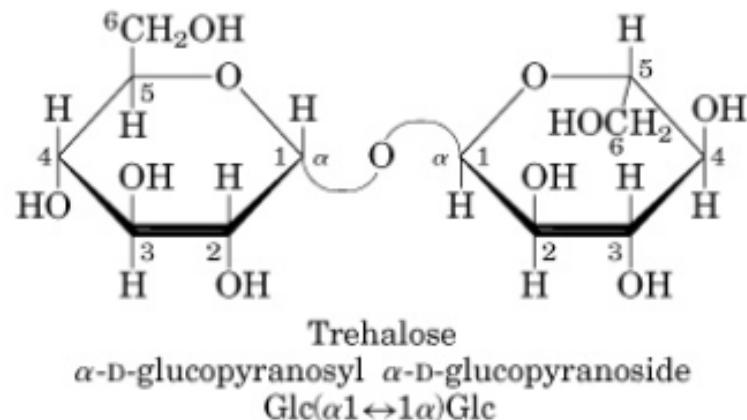
- The most abundant biomolecules on Earth
  - 100 billion tons/year of CO<sub>2</sub> & H<sub>2</sub>O into cellulose & other plant products via
- Most simple general formula:  
= C<sub>n</sub>(H<sub>2</sub>O)<sub>n</sub> where n ≥ 3
- Occur as:
  - monosaccharides - simple sugars (eg glucose)
  - oligosaccharides - 2 to 11 linked monosaccharides
    - disaccharides, tri, tetra, penta etc
  - polysaccharides - 12 to millions of linked monosaccharide units
- General name ending is often -ose:
  - glucose, lactose, cellulose BUT not always:
  - starch, glycogen, chitin

# Why carbohydrates are so important

- Energy Transport Molecules - Mono & Disaccharides
- Energy Storage - Polysaccharides
- Plant & Animal Structure - Polysaccharides
- Informational Molecules - Oligosaccharides

# Energy Transport Molecules

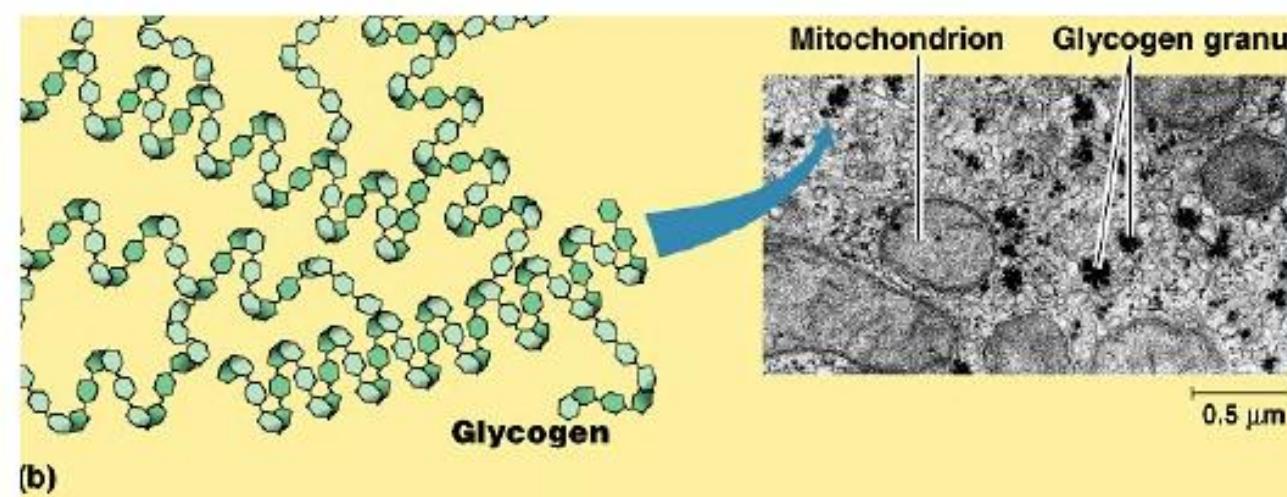
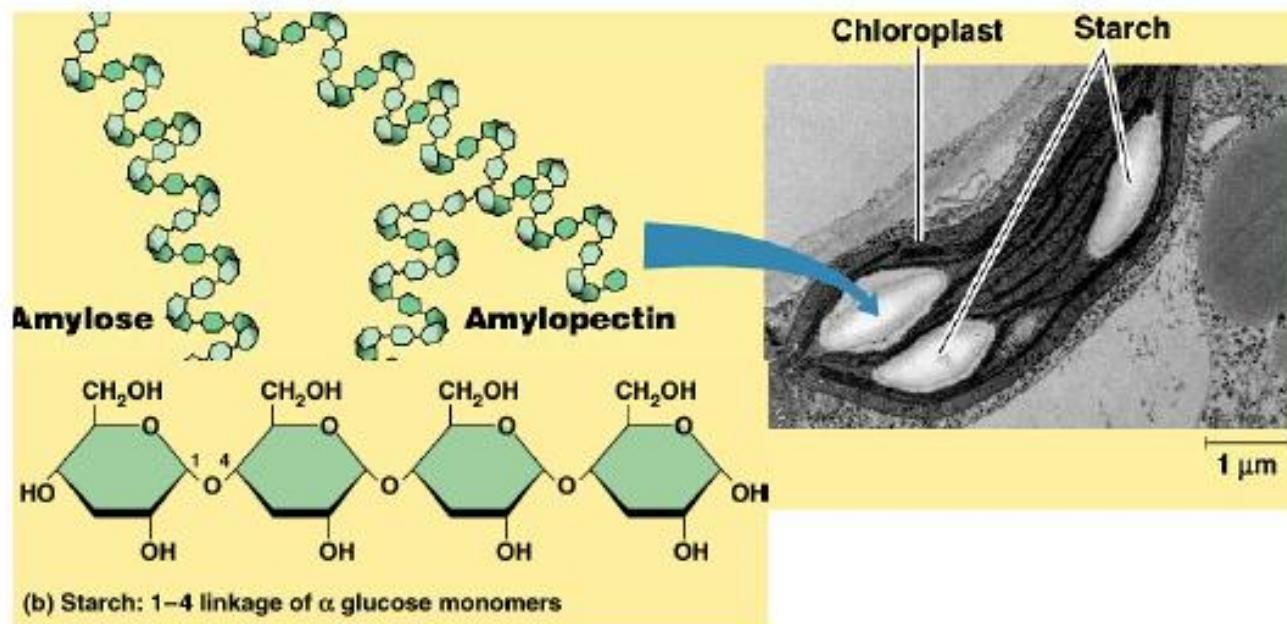
- Glucose: monosaccharide, in vertebrate blood
- Lactose:  $\beta$ -linked disaccharide, in mammalian milk
- Sucrose:  $\alpha\beta$ -linked disaccharide, in plants
- Trehalose:  $\alpha\alpha$ -linked glucose disaccharide, in insects



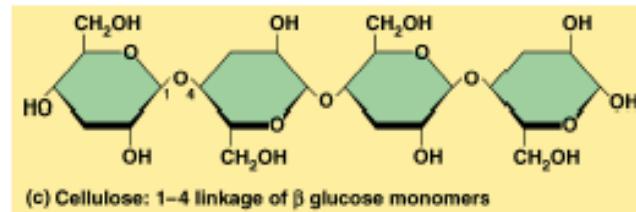
# Polysaccharides: Energy Storage or Structural

Homopolysaccharides		Heteropolysaccharides		
Unbranched	Branched	Two monomer types, unbranched	Multiple monomer types, branched	
				<ul style="list-style-type: none"><li>• Long polymeric chains of monosaccharides. Typically thousands to millions of residues in length</li><li>• Homopolysaccharides = one type of monosaccharide only</li></ul>
				<ul style="list-style-type: none"><li>• Heteropolysaccharides = more than one type (commonly 2)</li><li>• May be linear or branched</li><li>• May be neutral or may be negatively charged</li><li>• May be soluble, gum-like, granular or fibrous</li></ul>

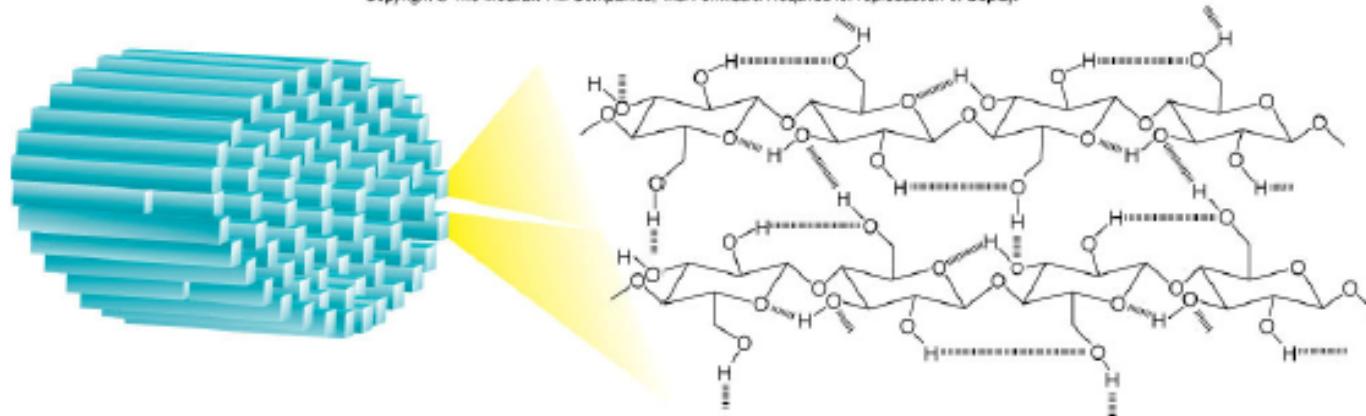
# Amylose : A Homopolysaccharide of Glucose



# Cellulose (Disaccharide is Cellobiose)



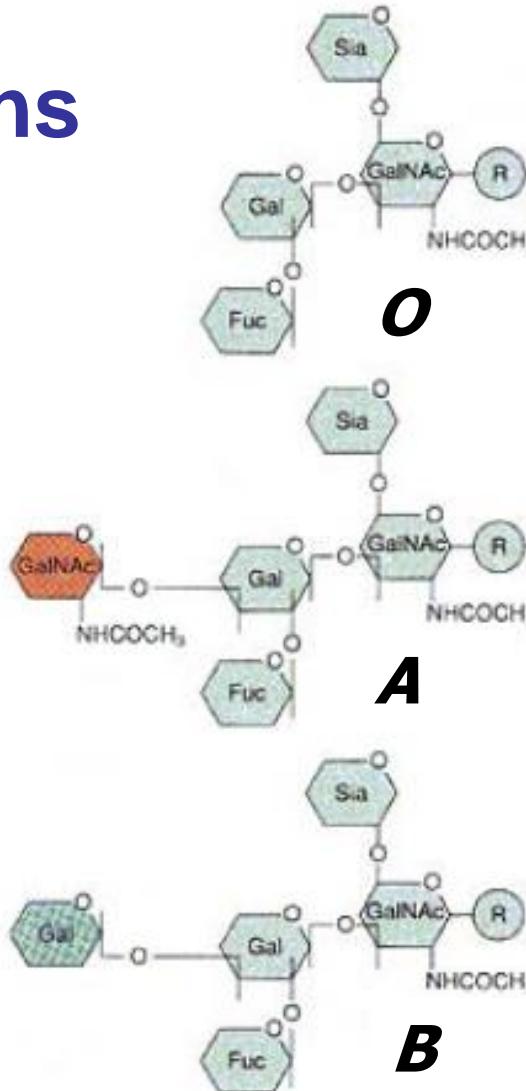
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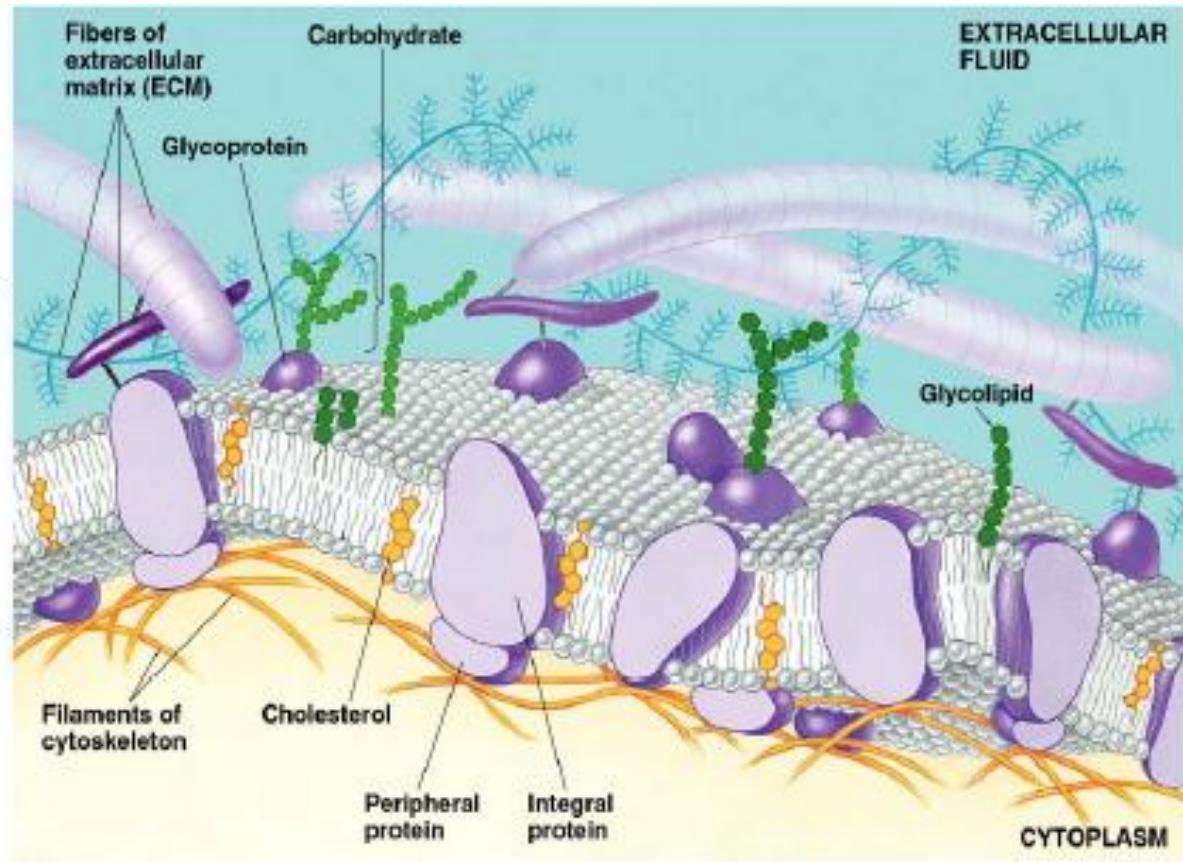
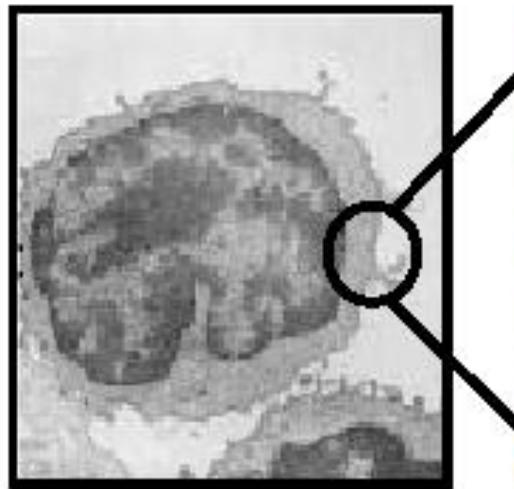
- Cellulose is the main structural component in wood, paper, cotton etc
- Most animals do not have an enzyme that attacks  $\beta(1 \rightarrow 4)$  O-glycosidic linkage in cellulose
- Herbivores carry bacteria that produce  $\beta(1 \rightarrow 4)$ cellulase
- Some termites, crabs etc may produce this enzyme

# Blood Group Antigens

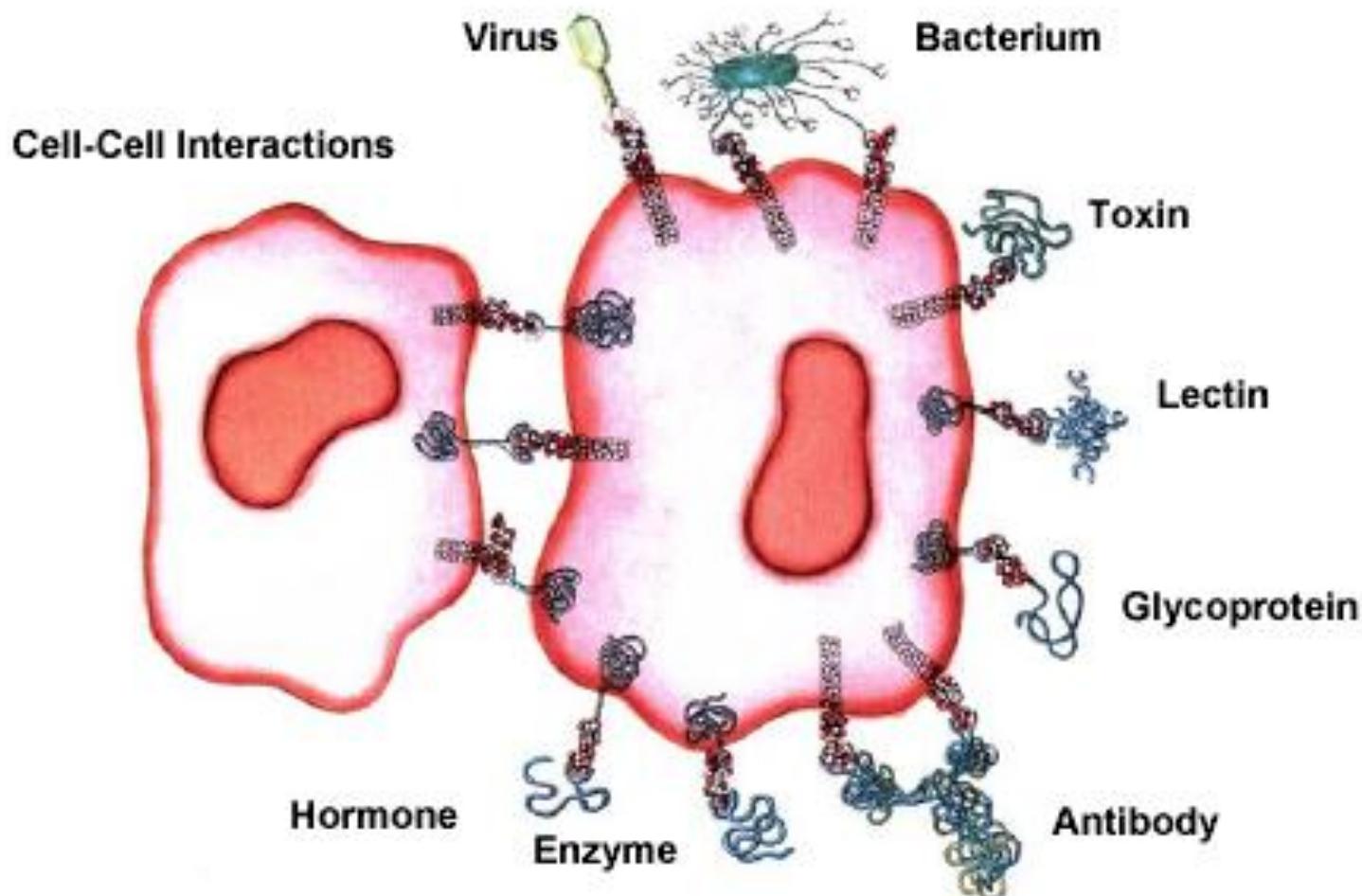
- ABO blood groups are determined by Oligosaccharides O-linked to proteins and lipids in the RBC membrane.
- Oligosaccharides typically carry additional groups eg amino sugars such as *N*-Acetylgalactosamine (GalNAc)
- Type A & B each elicit antigens that cause RBCs to “clump” together.



# Carbohydrates serve as recognition sites for other biomacromolecules



# Cell-cell interaction



# Lectin-carbohydrate interaction

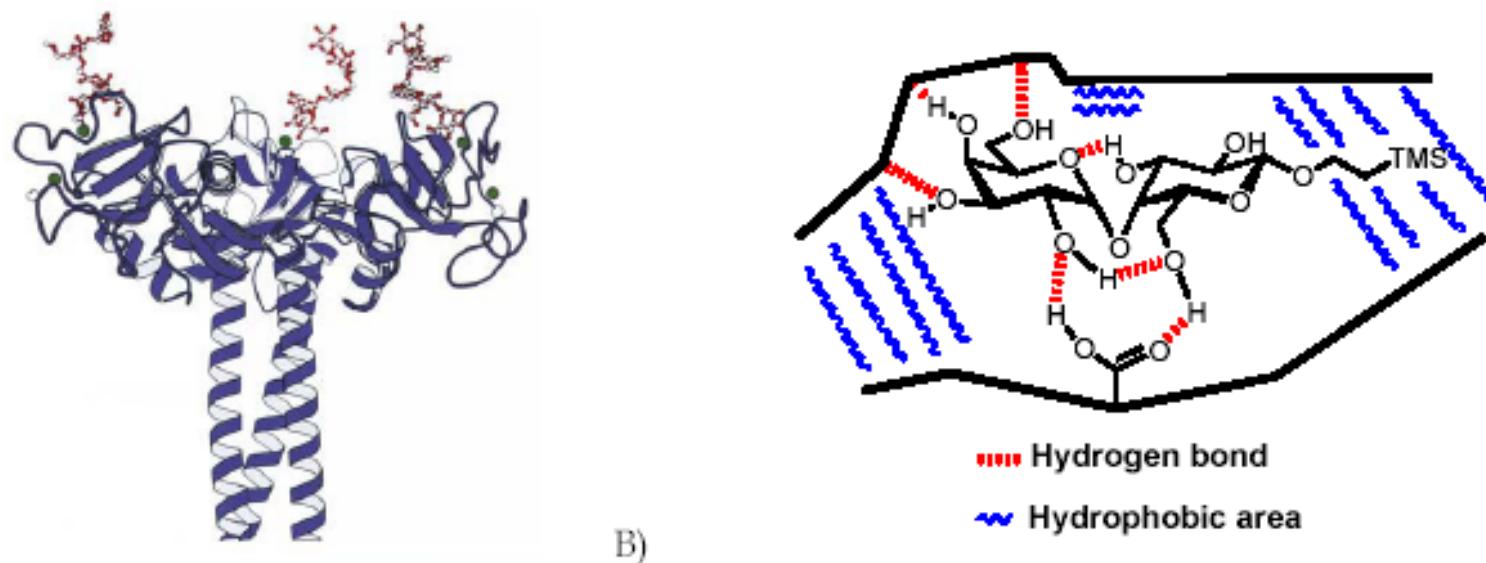
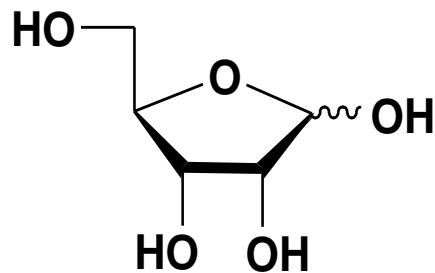
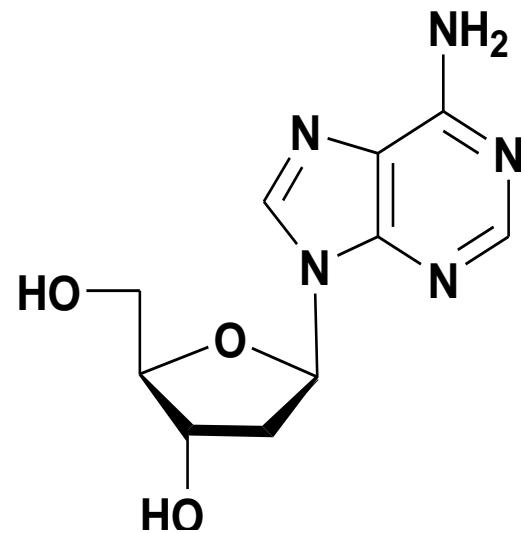


Fig. 1.11. Carbohydrate - lectin interactions. A) From Weis, W. I. and Drickamer, K. Trimeric structure of a C-type mannose-binding protein. *Structure* 1994, 2, 1227. B) Model of the galabiose binding site of PapGJ96.

# Carbohydrates are important for the storage of genetic information



D-ribose



Adenosine

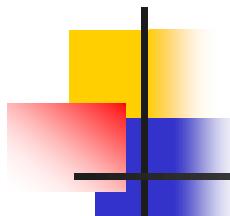


DNA

# Synthesis of Oligosaccharides



The challenge



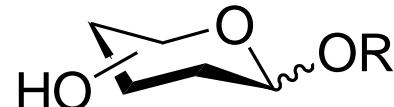
# Building Blocks

**MONO**



P = Hydroxyl Protection

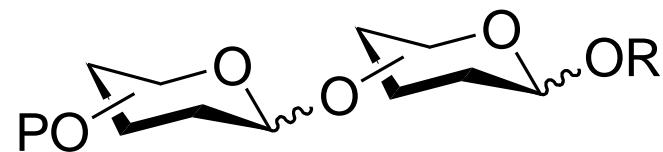
**MONO**



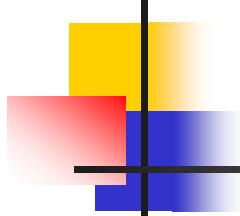
**DONOR**

Promoter

**ACCEPTOR**



**DI**



# DONORS



P = Hydroxyl Protection

Activating group

**Thioalkyl**

SMe, SEt, SPh, STol

**Halo**

F, Cl, Br, I

**Others:** *n*-Pentenyl

**Acetimides**

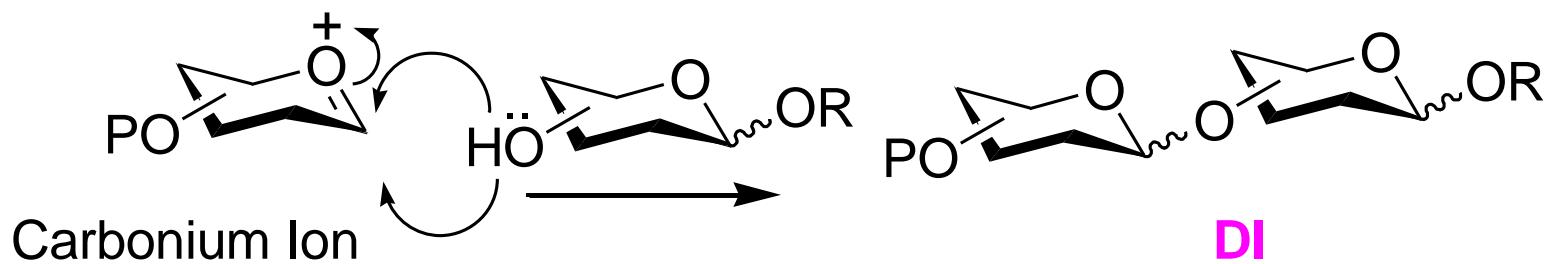
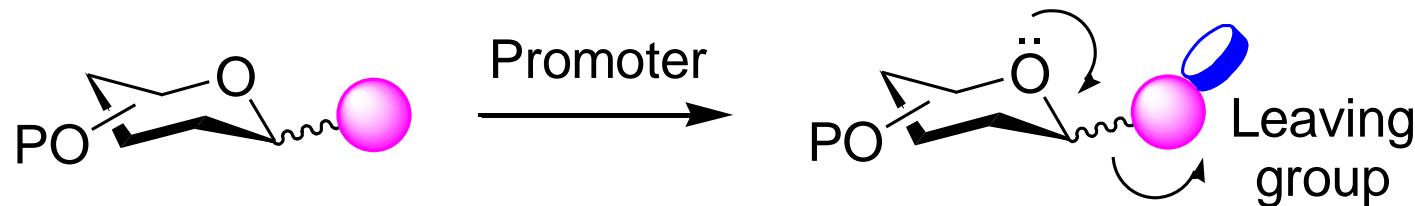
Trichloroacetimidate

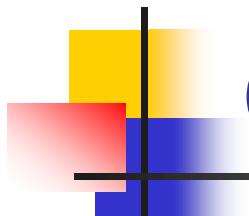
Trifluoroacetimidate

**Seleno**

SePh

# Glycosylation

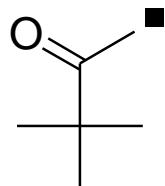




# Choice of hydroxyl protection

## Acyl protection

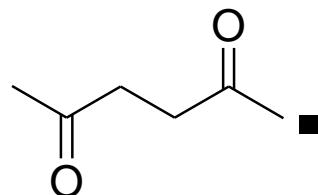
Acetate  
Benzoate  
Pivaloate  
Levulonate



Pivaloate

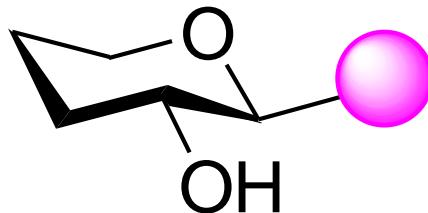
## Alkyl/aryl protection

Methyl  
Benzyl  
4-methoxybenzyl  
4-nitrobenzyl



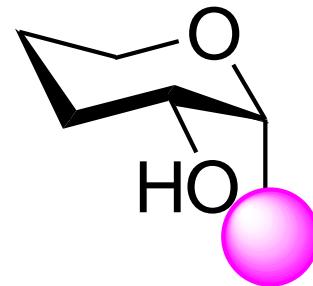
Levulonate

# Stereochemistry of Glycosylation



**1,2-*trans***

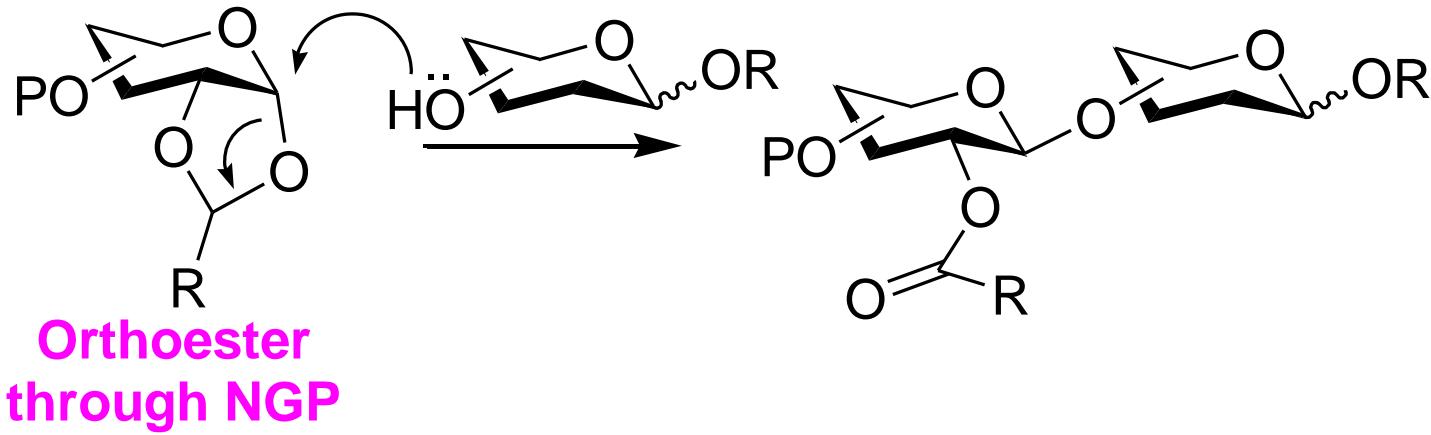
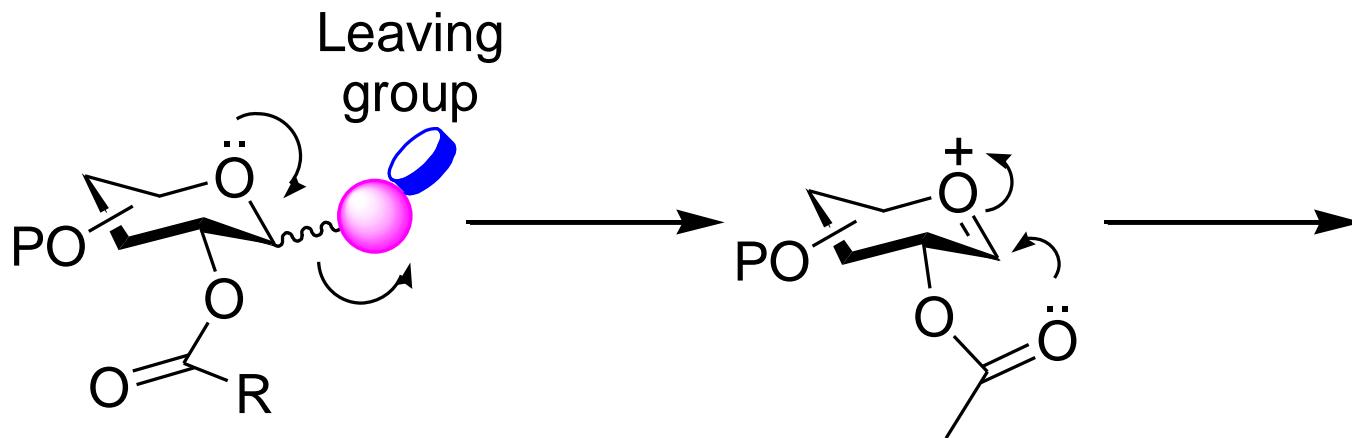
**Sterically favoured  
kinetic controlled**



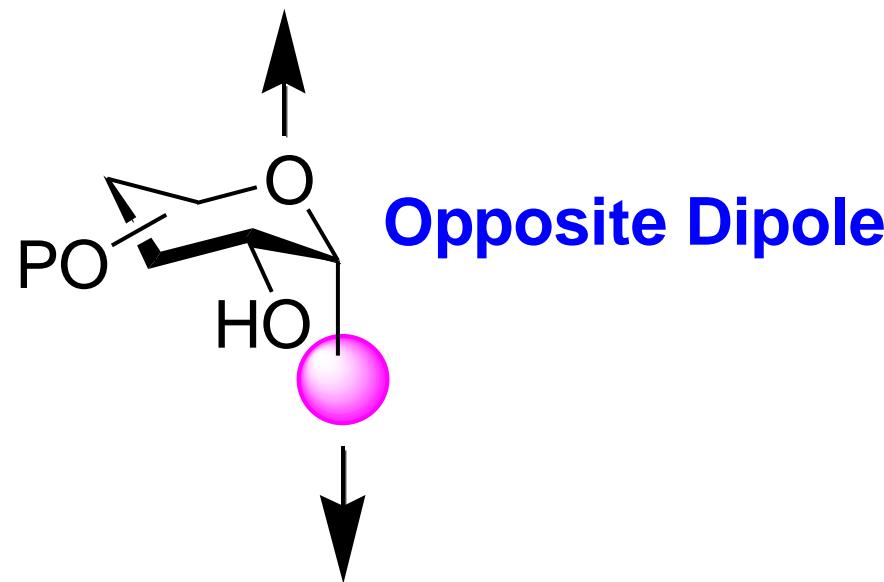
**1,2-*cis***

**Thermodynamic  
controlled**

# 1,2-trans Glycosylation

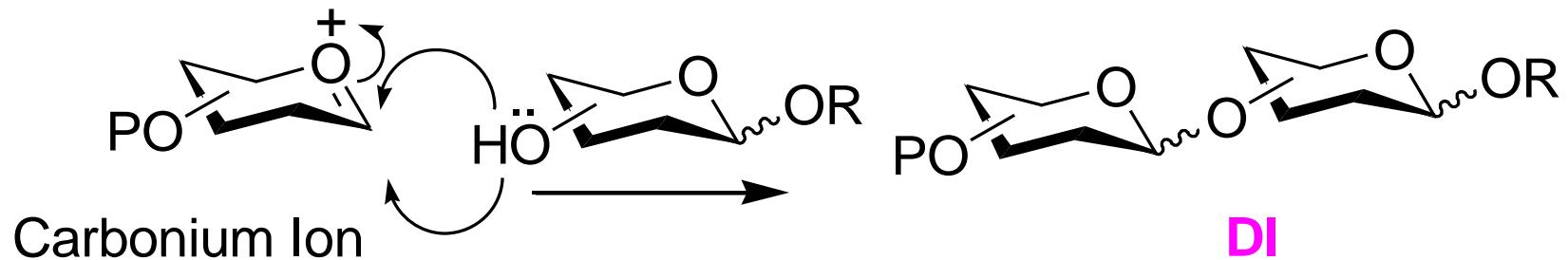


# 1,2-cis Glycosylation



Anomeric effect  
Thermodynamically controlled

# 1,2-cis Glycosylation



**CONDITIONS .....**

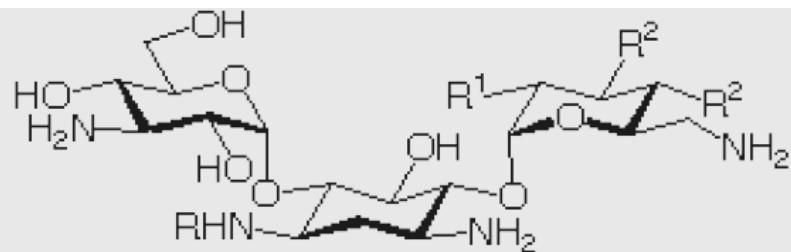
No participating group in 2-position

Low temperature

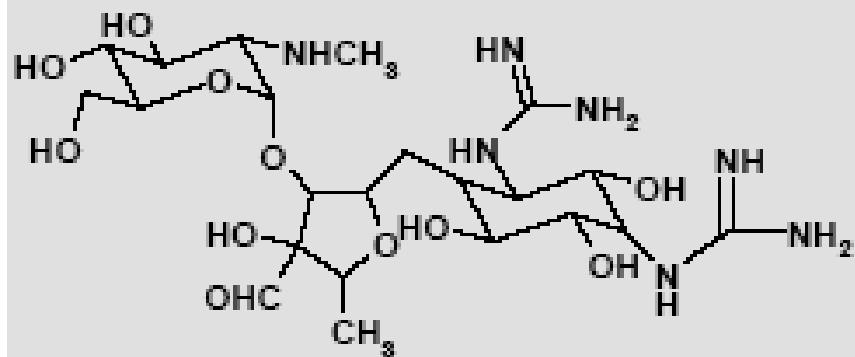
Non-polar solvent

# Promise of Carbohydrates

*Medicinal ???*



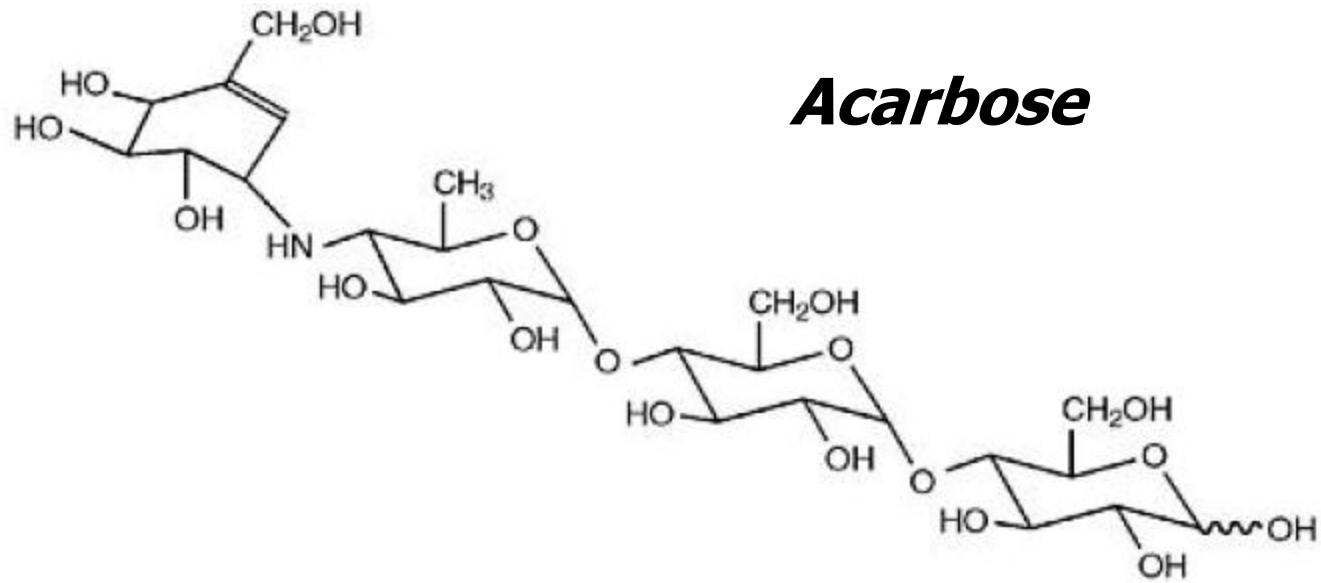
	R <sup>1</sup>	R <sup>2</sup>	R
Kanamycin	OH	OH	H
Kanamycin B	NH <sub>2</sub>	OH	H
Dibekacin	NH <sub>2</sub>	H	H
Arbekacin	NH <sub>2</sub>	H	COCH(OH)CH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub> (S)



1

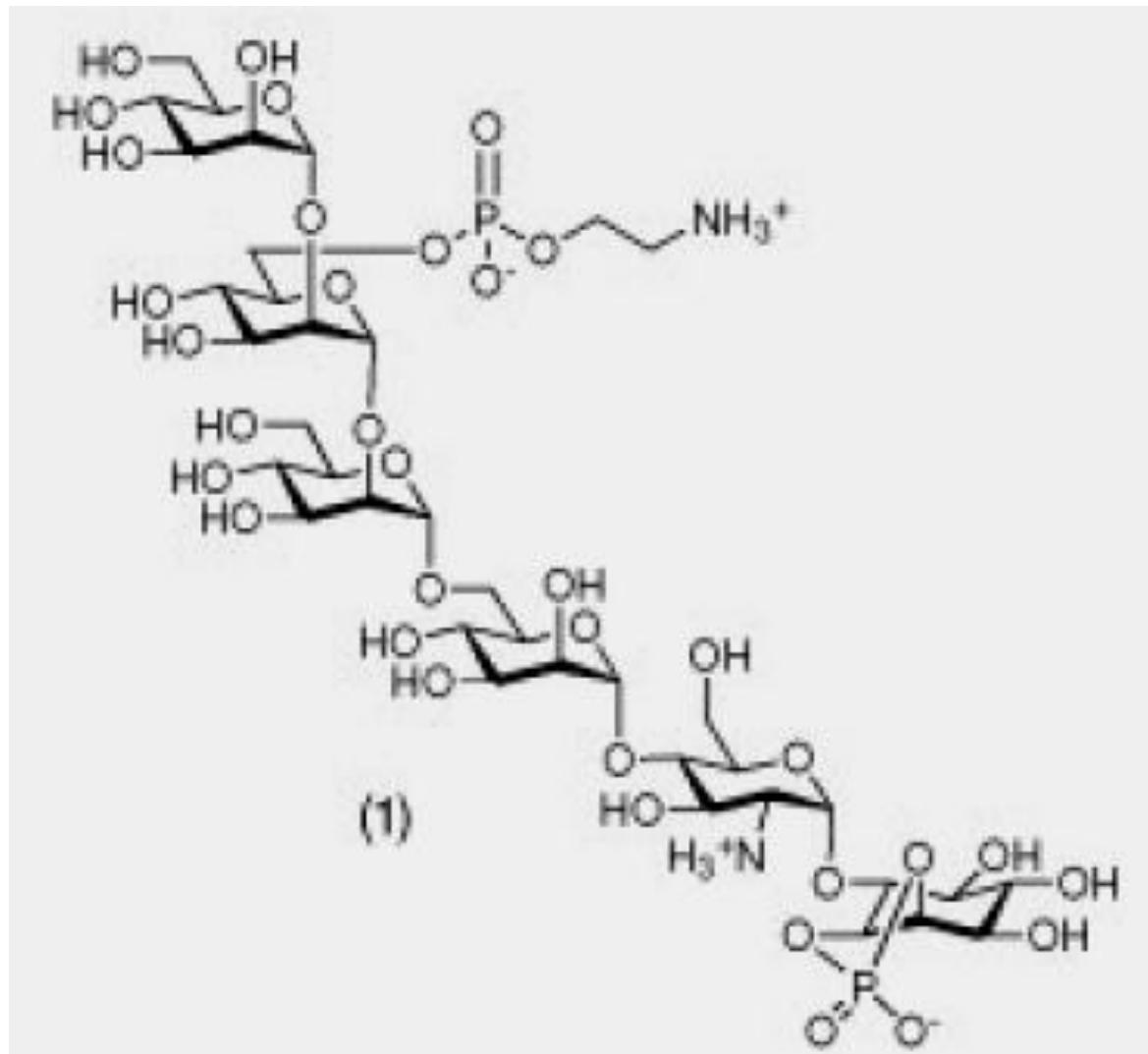
Streptomycin

***Anti tuberculosis***

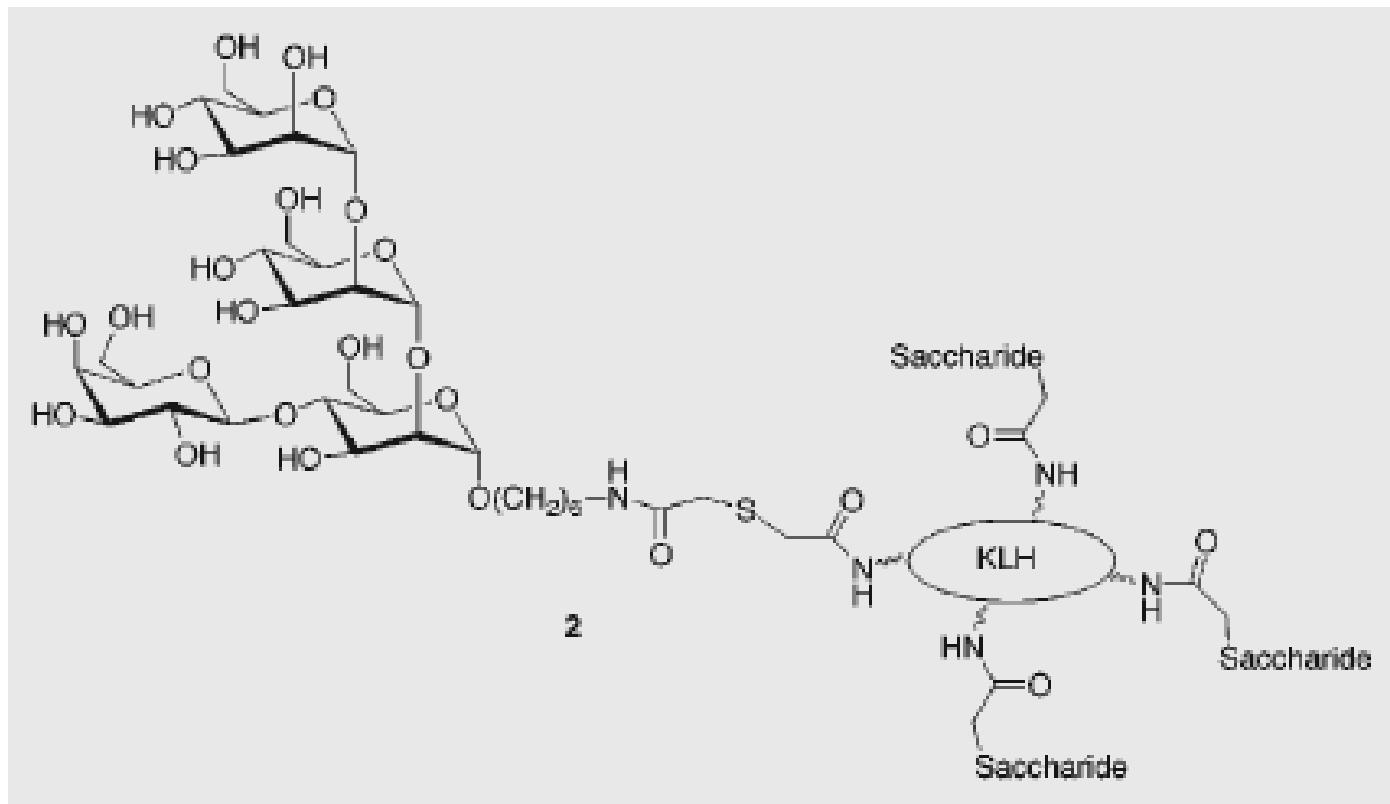


***Acarbose***

***Anti Diabetic***



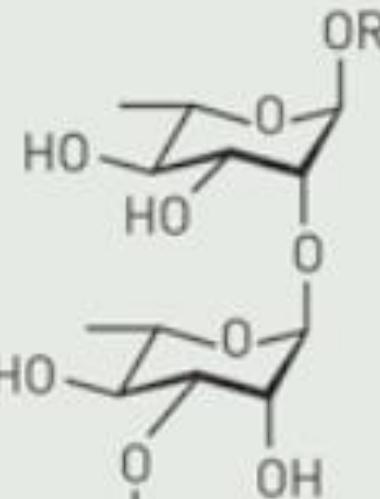
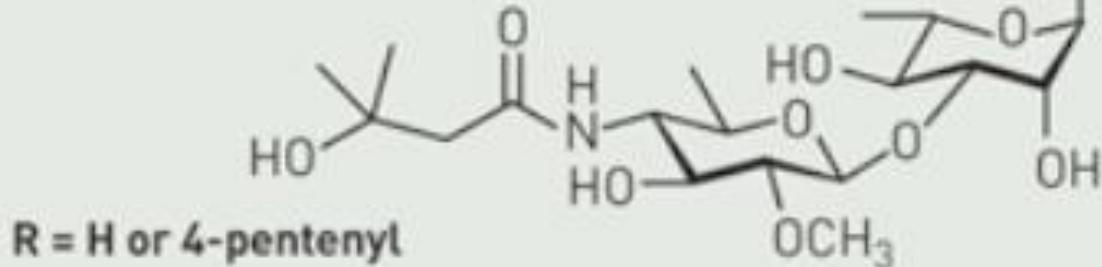
***Anti Malaria (Pete Seeberger)***



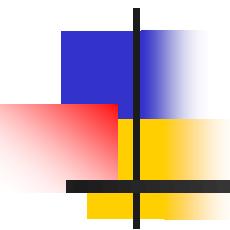
keyhole limpet hemocyanin (KLH)

***Anti Leishmania (Pete Seeberger)***

**VACCINE LEAD** Terminal tetrasaccharide of the *B. anthracis*' surface glycoprotein includes a unique monosaccharide, anthrose (bottom). A pentenyl group attached at top right facilitates conjugation to a carrier protein.



*Anthrax antigen (Pete Seeberger)*



# THANKS